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EIGHTH FLO			ART UNIT	PAPER NUMBER	
SAN FRANCISCO, CA 94111-3834			2814		

DATE MAILED: 04/21/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

r		Application No.	Applicant(s)	
Office Action Summary		09/747,779	KIM, HONG KOO	
		Examiner Art Unit		i
		Marcos D. Pizarro-Crespo	2814	pm
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A SHORTE THE MAILII - Extensions of after SIX (6) I - If the period f - If NO period f - Failure to rep Any reply rec	NED STATUTORY PERIOD FOR REPL NG DATE OF THIS COMMUNICATION. Itime may be available under the provisions of 37 CFR 1.1 MONTHS from the mailing date of this communication. or reply specified above is less than thirty (30) days, a rep or reply is specified above, the maximum statutory period by within the set or extended period for reply will, by statute eived by the Office later than three months after the mailin t term adjustment. See 37 CFR 1.704(b).	136(a). In no event, however, may a reply be ti ly within the statutory minimum of thirty (30) da will apply and will expire SIX (6) MONTHS fron e, cause the application to become ABANDONI	imely filed ys will be considered timely in the mailing date of this co ED (35 U.S.C.§ 133).	
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10)⊡ The de Applic Repla	pecification is objected to by the Examine rawing(s) filed on is/are: a) according and may not request that any objection to the cement drawing sheet(s) including the correctath or declaration is objected to by the Example.	cepted or b) objected to by the drawing(s) be held in abeyance. Setion is required if the drawing(s) is ob	ee 37 CFR 1.85(a). ojected to. See 37 CF	` '
Priority under	35 U.S.C. § 119			
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2) D Notice of Dra	erences Cited (PTO-892) Iftsperson's Patent Drawing Review (PTO-948) Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Mail Date	4) Interview Summary Paper No(s)/Mail D 5) Notice of Informal I 6) Other:	oate	-152)

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Attorney's Docket Number: 00939-073311 US

Filing Date: 12/22/2000

Claimed Priority Dates: 3/29/2000 (Provisional 60/193,046)

12/27/1999 (Provisional 60/173,175)

Applicant(s): Kim

Examiner: Marcos D. Pizarro-Crespo

DETAILED ACTION

This Office action responds to the amendment in paper no. 18 filed on 2/27/2004.

Acknowledgments

1. The amendment in paper no. 18, filed on 2/27/2004, responding to the Office action in paper no. 17, mailed on 10/21/2003, has been entered. The present Office action is made with all the suggested amendments being fully considered. Accordingly, pending in this Office action are claims 1-21, 23, 24, and 26-34.

Claim Rejections - 35 USC § 102

2. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

Claim Rejections - 35 USC § 103

3. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

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(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

- 4. Claims 1, 3-10, 14-18, 20, 21, 23, 24, and 27 are rejected under section 35 U.S.C. 103(a) as being unpatentable over Hirai (US 5955755), Kirlin (US 5225561), Maiti (US 6020024).
- 5. Regarding claim 1, Hirai shows (see, e.g., figs. 1, 2D, 2E, 3A, and 3B) most aspects of the instant invention including a method for fabricating a non-volatile memory device, the method comprising:
 - providing a silicon substrate 1
 - forming an oxide layer 4 on the substrate 1
 - forming a buffer layer 5 on the oxide layer 4 after forming the oxide layer 4 over the substrate 1, the buffer layer having a crystalline structure (see, e.g., col.3/II.53-57)
 - forming a ferroelectric material 6 overlying the substrate 1 and on the buffer layer 5
 - forming a gate layer 7 overlying the ferroelectric material 6 and a channel region
 - forming a first source/drain region 2 adjacent to a first side of the channel region
 - forming a second source/drain region 3 adjacent to a second side of the channel region

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Hirai, however, fails to show a step of thermally annealing the buffer layer. Nonetheless, annealing steps are thermal treatments commonly used in the semiconductor art to eliminate various weaknesses, or to produce other qualities in a material. As taught by Kirlin (see, e.g., col.35/II.30-33) and Maiti (see, e.g., col.3/II.33-36 and col.4/II.10-14), such an annealing step will remove the vacancies of Hirai's buffer layer, thereby improving its quality and reducing its defects.

It would have been obvious at the time of the invention to one of ordinary skill in the art to include in Hirai's method a step of thermally annealing the buffer layer, as suggested by Kirlin and Maiti, to improve the quality of the layer.

With respect to the gate oxide, although Hirai fails to specify that the oxide layer is amorphous, he (see, e.g., col.7/II.38) shows that the layer is a silicon-dioxide layer. The recited amorphicity of Hirai's silicon dioxide layer is inherent to Hirai's oxide layer. See, e.g., pp.7/II.7-8 of the instant specification.

- 6. Regarding claim 3, Hirai shows that the ferroelectric material may be a PZT-bearing compound (see, e.g., col.2/II.40-43).
- 7. Regarding claim 4, Hirai shows that the buffer layer may be a magnesium-bearing compound (see, e.g., col.2/II.39).
- 8. Regarding claim 5, Hirai shows that the buffer layer may be a magnesium-oxide layer, the magnesium-oxide layer being a barrier layer (see, e.g., col.2/II.39, col.3/II.7-10, col.4/II.62-67).
- 9. Regarding claim 6, Hirai shows that the ferroelectric material may have a thickness of less than about 1000 angstroms (see, *e.g.*, col.4/II.42).

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- 10. Regarding claim 7, Hirai shows that the buffer layer may have a thickness of 20 nanometers (see, e.g., col.4/II.13-14).
- 11. Regarding claim 8, Hirai shows that the ferroelectric material may have a thickness greater than 100 angstroms (see, *e.g.*, col.4/II.42).
- 12. Regarding claim 9, Hirai shows that the ferroelectric material may be PZT (see, e.g., col.4/II.37).
- 13. Regarding claim 10, Hirai shows that the buffer layer is a diffusion barrier layer substantially preventing the diffusion between the ferroelectric material and the substrate (see, e.g., col.4/II.62-67).
- 14. Regarding claim 14, Hirai shows that the ferroelectric material is highly oriented (see, e.g., col.4/II.40-41, col.5/II.48-52).
- 15. Regarding claims 15 and 16, Hirai (see, *e.g.*, col.10/ll.17-18) shows that the ferroelectric material is a highly oriented (001) thin film. The polycrystallinity of Hirai's highly oriented ferroelectric material is an inherent property. Hirai, for example, shows that his ferroelectric material is a highly oriented PZT thin-film showing a (001)-face (see, *e.g.*, col.4/ll.50-52). (001)-PZT planes are polycrystalline (see remarks section below).
- 16. Regarding claim 17, Hirai shows that the polycrystalline film is greater than 100 angstroms (see, e.g., col.4/II.42).
- 17. Regarding claim 18, Hirai shows that the buffer layer is a template to provide an oriented growth of the ferroelectric film (see, e.g., col.4/II.31-35)

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- 18. Regarding claim 20, Hirai shows that the oxide layer is silicon dioxide (see, *e.g.*, col.9/II.58). The substrate-surface passivation-property is inherent to Hirai's oxide layer (see remarks section below).
- 19. Regarding claim 21, Hirai shows (see, *e.g.*, figs. 1, 2D, 2E, 3A, and 3B) most aspects of the instant invention including a method for fabricating a non-volatile memory device, the method comprising:
 - providing a semiconductor substrate 1
 - > forming a first buffer layer as a gate oxide layer 4 on the substrate 1
 - ▶ forming a second buffer layer as a MgO layer 5 on the oxide layer 4 after forming the oxide layer 4 on the substrate 1, the MgO layer having a crystal structure (see, e.g., col.3/II.53-57)
 - > forming a ferroelectric material 6 overlying the substrate and the MgO layer 5
 - forming a gate layer 7 overlying the ferroelectric material 6 and a channel region
 - forming first and second doped regions 2, 3 adjacent to first and second ends of the channel region

Hirai, however, fails to show a step of thermally annealing the buffer layer. Nonetheless, annealing steps are thermal treatments commonly used in the semiconductor art to eliminate various weaknesses, or to produce other qualities in a material. For example, Kirlin (see, e.g., col.35/II.30-33) and Maiti (see, e.g., col.3/II.33-36 and col.4/II.10-14) teaches that thermally annealing Hirai's buffer layer will remove the vacancies of the layer, thereby improving its quality and reducing its defects.

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It would have been obvious at the time of the invention to one of ordinary skill in the art to include in Hirai's method a step of thermally annealing the buffer layer, as suggested by Kirlin and Maiti, to improve the quality of the layer.

With respect to the gate oxide, Hirai fails to specify that the oxide layer is non-crystalline. He (see, e.g., col.7/II.38), however, shows that the layer is a silicon-dioxide layer. The non-crystallinity of Hirai's silicon dioxide layer is an inherent property. See, e.g., pp.7/II.7-8, where the applicant teaches that silicon dioxides are non-crystalline materials.

- 20. Regarding claim 23, Hirai (see, e.g., col.4/II.49) shows that the MgO layer is a highly oriented layer, whereas the applicant teaches (see, e.g., pp.7/II.7-8) that silicon dioxide are amorphous.
- 21. Regarding claim 24, although Hirai does not show the MgO layer has a thickness of less than 10 nm, he shows that his paraelectric oxide thin layer may be approximately 14 nm (see, e.g., col.12/II.3). Hirai's thickness appears to be closed enough to the claimed thickness range that one of ordinary skill in the art would have expected Hirai's MgO layer to have the same properties as those of the claimed layer; consequently, it would have been obvious. *Titanium Metal Corp. of America v. Banner*, 778 F.2d 775, 227 USPQ 773 (Fed. Cir. 1985).

Moreover, it would be an obvious matter of design choice to have a thickness not greater than 10 nm, instead of 14 nm, for Hirai's MgO layer since such a modification would have involved a mere change in the size of the layer. A change in size is

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generally recognized as being within the level of ordinary skill in the art. *In re Rose*, 105 USPQ 237 (CCPA 1955).

- 22. Regarding claim 27, Hirai shows that the MgO layer has a highly oriented structure. Kirlin (see, e.g., col.35/ll.1-33) shows that the MgO layer had a highly oriented structure prior to the annealing step.
- 23. Claim 2 is rejected under 35 U.S.C. 103(a) as being unpatentable over Hirai/Kirlin/Maiti in view of Yamazaki (US 6072724).
- 24. Regarding claim 2, Hirai/Kirlin/Maiti shows most aspects of the instant invention (see paragraphs 5-22 above) but fails to specify the channel region to be about 1 micron or less.

Yamazaki (see, e.g., col.2/II.45-46), on the other hand, teaches that it is known that the channel length is an important design parameter that will determine the channel current of the transistor.

Consequently, it would be an obvious matter of design choice to select a suitable channel length for the transistor of Hirai/Kirlin/Maiti, as suggested by Yamazaki, since the channel length is a variable of importance subject to routine experimentation and optimization and it is not inventive to discover the optimum or workable ranges by routine experimentation.

- 25. Claims 11 and 12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hirai/Kirlin/Maiti in view of Van Zant and Evetts (US 5361720).
- 26. Regarding claim 11, Hirai/Kirlin/Maiti shows most aspects of the instant invention (see paragraphs 5-22 above), except for sputtering the buffer material from a

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substantially pure magnesium target to form a magnesium oxide layer. Hirai differently deposits the magnesium layer using vacuum evaporation (see, e.g., col.4/ll.10-12).

Van Zant (pp.412), however, teaches that there are several advantages to the use of sputtering over vacuum evaporation. One is the improvement in step coverage. Evetts (see, e.g., col.2/II.15-20), on the other hand, teaches that sputter deposition from a magnesium metal target in a sputtering gas comprising oxygen is a preferred method of forming a high-quality MgO layer.

Consequently, it would have been obvious at the time of the invention to one of ordinary skill in the art to sputter Hirai/Kirlin/Maiti/lyer's buffer layer from a magnesium target to form the MgO layer, as suggested by Van Zant and Evetts, in order to improve the step coverage while forming a high-quality MgO layer.

27. Regarding claim 12, Hirai/Kirlin/Maiti/Van Zant/Evetts shows most aspects of the instant invention (see paragraph 26 above). In addition, Evetts (see, e.g., col.2/II.2) shows that the deposition temperature may be as low as 540°C. Hirai/Kirlin/Maiti/Van Zant/Evetts, however, fails to teach a sputtering temperature between 400-500°C.

In spite of the above, generally, differences in temperature will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical. "Where the general conditions of a claim are disclosed in the prior art, it is not inventive to discover the workable ranges by routine experimentation". *In re Aller*, 220 F.2d 454,456,105 USPQ 233, 235 (CCPA 1955).

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There is no evidence supporting the claimed temperature range, *i.e.*, 400-500°C, as critical to the invention, and therefore it would have been obvious.

- 28. Claim 13 is rejected under 35 U.S.C. 103(a) as being unpatentable over Hirai/Kirlin/Maiti/Van Zant/Evetts and Wolf.
- 29. Hirai/Kirlin/Maiti/Van Zant/Evetts shows most aspects of the instant invention (see paragraph 26 above). In addition, Kirlin (see, e.g., col.35/II.30-35) shows that the annealing step may be performed at 700°C for 20 minutes. Hirai/Kirlin/Maiti/Van Zant/Evetts, however, fails to perform the annealing between 800-1000°C for about 30 minutes.

Wolf (pp.57), on the other hand, teaches that it is known that temperature and time are important design parameters affecting wafer warpage or slip and dopant diffusion.

Consequently, it would be an obvious matter of design choice to select a suitable temperature and time for the anneal step of Hirai/Kirlin/Maiti/Van Zant/Evetts, as taught by Wolf, since these are variables of importance subject to routine experimentation and optimization and it is not inventive to discover the optimum or workable ranges by routine experimentation.

In spite of the above, generally, differences in temperature and time will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical. "Where the general conditions of a claim are disclosed in the prior art, it is not inventive to discover the

workable ranges by routine experimentation". *In re Aller*, 220 F.2d 454,456,105 USPQ 233, 235 (CCPA 1955).

There is no evidence supporting the claimed temperature range and time value, as critical to the invention, and therefore they would have been obvious.

- 30. Claim 19 is rejected under 35 U.S.C. 103(a) as being unpatentable over Hirai/Kirlin/Maiti in view of Jaeger.
- 31. Regarding claim 19, Hirai/Kirlin/Maiti shows most aspects of the instant invention (see paragraphs 5-22 above). Hirai (see, e.g., col.8/II.2) also teaches that the oxide layer is formed by a thermal oxidation step. He, however, fails to teach the use of a dry oxidation process to form the oxide layer. Jaeger (see, e.g., pp.42), on the other hand, teaches that dry oxidation would have been commonly used as the thermal oxidation step of Hirai to maintain good process control.

It would have been obvious at the time of the invention to one of ordinary skill in the art to use dry oxidation as the thermal oxidation step in Hirai/Kirlin/Maiti's method, as suggested by Jaeger, to maintain good process control.

- 32. Claim 26 is rejected under 35 U.S.C. 103(a) as being unpatentable over Hirai/Kirlin/Maiti in view of Wolf.
- 33. Regarding claim 26, see the comments stated above in paragraph 29 with respect to claim 13, which are considered repeated here.
- 34. Claims 29 and 31 are rejected under section 35 U.S.C. 102(e) as anticipated by Hirai or, in the alternative, under 35 U.S.C. 103(a) as obvious over Hirai.

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- 35. Regarding claim 29, Hirai shows (see, *e.g.*, figs. 1, 2D, 2E, 3A, and 3B) all/most aspects of the instant invention including a method for fabricating a non-volatile memory device, the method comprising:
 - providing a semiconductor substrate 1
 - forming an oxide layer 4 on the substrate 1
 - forming a highly oriented MgO layer 5 on the oxide layer 4
 - forming a ferroelectric material 6 overlying the substrate 1 and on the MgO layer 5
 - forming a gate layer 7 overlying the ferroelectric material 6 and a channel region
 - forming first and second doped regions 2, 3 adjacent to the first and second ends of the channel region

As to the grounds of rejection under section 103(a), the non-crystallinity is inherent to Hirai's oxide layer. Although Hirai fails to specify that the oxide layer is non-crystalline, he shows that the layer is a silicon dioxide layer (see, e.g., col.7/II.38). The non-crystallinity of Hirai's oxide layer is inherent property to the layer. See, e.g., pp.7/II.7-8 of the instant specification, where the applicant teaches that silicon dioxides are non-crystalline materials.

See also MPEP § 2112, which discusses the requirements of rejections based on an inherent property and recommends the alternative (§ 102/ § 103) grounds of rejection.

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- 36. Regarding claim 31, Hirai shows (see, *e.g.*, figs. 3A and 3B) that the MgO layer **5** is formed after the oxide layer **4** is formed.
- 37. Claims 32-33 are rejected under section 35 U.S.C. 103(a) as being unpatentable over Hirai and Iyer.
- 38. Regarding claim 32, Hirai shows (see, e.g., figs. 1, 2D, 2E, 3A, and 3B) most aspects of the instant invention including a method for fabricating a non-volatile memory device, the method comprising:
 - providing a semiconductor substrate 1
 - forming a gate dielectric layer 4 on the substrate 1
 - ▶ forming an MgO layer 5 on the dielectric layer 4 after forming the dielectric layer 4 on the substrate 1, the MgO layer 5 having a highly-oriented structure
 - → forming a ferroelectric layer 6 overlying the MgO layer 5 (see, e.g., col.3/II.53-col.4/II.12)

wherein the dielectric layer **4**, the MgO layer **5**, and the ferroelectric layer **6** are patterned to form a transistor.

Hirai, however, fails to specify that the gate dielectric layer is amorphous. Nonetheless, Hirai (see, e.g., col.7/II.38) shows that the gate dielectric layer is a silicondioxide layer used as a gate dielectric in a memory device. As taught by Iyer (see, e.g., col.1/II.16-20), silicon dioxides commonly used as dielectrics in integrated circuits are amorphous materials.

Consequently, it would have been obvious at the time of the invention to one of ordinary skill in the art at the time of the invention that Hirai's gate dielectric layer is

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amorphous, as taught by lyer, since this is a silicon dioxide layer used as a dielectric and silicon dioxides commonly used as dielectrics in integrated circuits are amorphous.

- 39. Regarding claim 33, Hirai shows that the MgO layer has a crystalline structure (see, e.g., col.3/II.53-col.4/II.12).
- 40. Claim 30 is rejected under 35 U.S.C. 103(a) as being unpatentable over Hirai in view of Kirlin and Maiti.
- 41. Regarding claim 30, Hirai shows most aspects of the instant invention (see paragraphs 35-36 above), except for the step of thermally annealing the MgO layer. Nonetheless, annealing steps are thermal treatments commonly used in the semiconductor art to eliminate various weaknesses, or to produce other qualities in a material. For example, Maiti (see, e.g., col.3/II.33-36 and col.4/II.10-14) teaches that annealing Hirai's MgO layer will remove the vacancies of the layer, thereby improving its quality and reducing its defects. Likewise, Kirlin (see, e.g., col.35/II.7,30-33) teaches that annealing a highly oriented MgO layer eliminates oxygen anion vacancies within the layer.

It would have been obvious at the time of the invention to one of ordinary skill in the art to include in Hirai method a step of thermally annealing the MgO layer, as suggested by Kirlin and Maiti, to improve the quality of the layer.

Allowable Subject Matter

42. Claims 28 and 34 are objected to as being dependent upon a rejected base claim, but would be allowable if rewritten in independent form including all of the limitations of the base claim and any intervening claims.

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Remarks

43. As Auciello teaches (US 5453661/col.5/II.7-25), the highly oriented ferroelectric film of Hirai is polycrystalline.

44. As Van Zant teaches (pp.154-155), silicon dioxide layers are characterized by its passivating effect over the surface of a substrate.

Response to Arguments

45. The applicants argue:

The claimed invention is clearly different from Hirai. Hirai differently forms *first* the oriented buffer layer on the substrate and *then* the silicon oxide layer is formed by an anneal step in an oxygen ambient so as to be sandwiched between the substrate and the buffer layer. This resulted in a lower quality silicon oxide layer, which indicates that Hirai did not know how to form an oriented buffer layer on a non-crystalline surface.

The examiner responds:

Contradicting applicants' statement, Hirai clearly shows a process in which the silicon oxide layer is formed *first* and *then* the oriented buffer layer is formed on the oxide layer (see, e.g., Hirai/col.7/II.66-col.8/II.10).

46. The applicants argue:

The process that Hirai uses to form the silicon oxide buffer is very different from the process of the instant invention. Therefore, the properties of the resulting interface between the silicon oxide and the silicon substrate are expected to be different. The nature of Hirai's silicon surface passivation cannot be the same as that of the present invention, which employs a thermal oxidation of a free silicon surface without any prior material deposited thereon.

The examiner responds:

Hirai clearly teaches (see, *e.g.*, col.7/II.66-col.8/II.10) to form a silicon oxide by thermal oxidation of a free silicon surface without any prior material deposited thereon. According to the applicants, since Hirai and the claimed invention use the same process to form a silicon oxide layer on a silicon surface, the passivating properties of Hirai's silicon oxide are expected to be the same as those of the claimed invention.

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Conclusion

47. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

- 48. A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.
- 49. Papers related to this application may be submitted directly to Art Unit 2814 by facsimile transmission. Papers should be faxed to Art Unit 2814 via the Art Unit 2814 Fax Center. The faxing of such papers must conform to the notice published in the Official Gazette, 1096 OG 30 (15 November 1989). The Art Unit 2814 Fax Center number is (703) 872-9306. The Art Unit 2814 Fax Center is to be used only for papers related to Art Unit 2814 applications.
- 50. Any inquiry concerning this communication or earlier communications from the examiner should be directed to **Marcos D. Pizarro-Crespo** at **(571) 272-1716** and between the hours of 9:00 AM to 7:30 PM (Eastern Standard Time) Monday through

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Thursday or by e-mail via Marcos.Pizarro@uspto.gov. If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Wael Fahmy, can be reached on (571) 272-1705.

- 51. Any inquiry of a general nature or relating to the status of this application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).
- 52. The following list is the Examiner's field of search for the present Office Action:

Field of Search	Date
U.S. Class / Subclass(es): 257/295, 438/3, 365/145	4/13/2004
Other Documentation: PLUS Analysis	9/11/2002
Electronic Database(s): EAST (USPAT, EPO, JPO)	4/13/2004

Lelfalm SOE 284

Marcos D. Pizarro-Crespo Patent Examiner Art Unit 2814

Art Unit 2814 571-272-1716

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